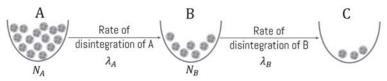
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RADIOACTIVE SERIES

Radioactive equilibrium



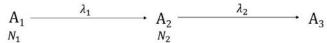
Rate of disintegration of element A, $R_{A}=\lambda_{A}N_{A}$

Rate of disintegration of element B. $R_B = \lambda_B N_B$

Presently, the speed at which element A breaks down matches the speed at which element B forms. When the speed of formation matches the speed of breakdown for element B, it's termed as radioactive equilibrium.

Radioactive series

Accumulation of a Radioactive Element in Radioactive Series



Let at time t = 0, N_0 be the number of active nuclei of A_1 and no. of active nuclei of A_2 is zero. The no. of active nuclei of A_2 at time t is

 $N_1 = N_0 e^{-\lambda_1 t}$

Decay rate of $A_1 = Production rate of A_2 = \lambda_1 N_1$

Decay rate of $A_2 = \lambda_2 N_2$

Accumulation rate of A2

$$\frac{\mathrm{dN_2}}{\mathrm{dt}} = \lambda_1 \mathrm{N_1} - \lambda_2 \mathrm{N_2} \qquad \qquad \begin{aligned} \lambda_1 &= \text{ Decay constant of } A_1 \\ \lambda_2 &= \text{ Decay constant of } A_2 \end{aligned}$$

$$\begin{split} \frac{dN_2}{dt} &= \lambda_1 N_1 - \lambda_2 N_2 \\ dN_2 &+ \lambda_2 N_2 at = \lambda_1 N_1 dt \\ dN_2 &+ \lambda_2 N_2 dt = \lambda_1 N_0 e^{-\lambda_1 t} dt \left(\because N_1 = N_0 e^{-\lambda_1 t} \right) \end{split}$$

Multiplying both sides by $e^{\lambda_2 t}$

$$\begin{split} dN_2 e^{\lambda_2 t} + \lambda_2 e^{\lambda_2 t} N_2 dt &= \lambda_1 N_0 e^{(\lambda_2 - \lambda_1) t} dt \\ d(N_2 e^{\lambda_2 t}) &= \int \lambda_1 N_0 e^{(\lambda_2 - \lambda_1) t} dt \\ N_2 e^{\lambda_2 t} &= \lambda_1 N_0 \frac{e^{(\lambda_2 - \lambda_1) t}}{\lambda_2 - \lambda_1} + C \end{split}$$

at t = 0; $N_2 = 0$

$$\begin{split} 0 &= \frac{\lambda_1 N_0}{\lambda_2 - \lambda_1} + c \\ C &= \frac{\lambda_1 \lambda_0}{\lambda_1 - \lambda_2} \\ N_2 e^{\lambda_2 t} &= \frac{\lambda_1 N_0}{\lambda_1 - \lambda_2} e^{(\lambda_2 - \lambda_1) t} \frac{+ \lambda_1 N_0}{\lambda_1 - \lambda_2} \\ N_2 e^{\lambda_2 t} &= \frac{\lambda_1 N_0}{\lambda_1 - \lambda_2} \left[1 - e^{(\lambda_2 - \lambda_1) t} \right] \\ N_2 &= \frac{\lambda_1 N_0}{\lambda_1 - \lambda_2} \left[\frac{1}{e^{\lambda_2 t}} - \frac{e^{(\lambda_2 - \lambda_1) t}}{e^{\lambda_2 t}} \right] \\ N_2 &= \frac{\lambda_1 N_0}{\lambda_1 - \lambda_2} \left[e^{-\lambda_2 t} - e^{(\lambda_2 - \lambda_1 - \lambda_2) t} \right] \\ N_2 &= \frac{N_0 \lambda_1}{\lambda_1 - \lambda_2} \left[e^{-\lambda_2 t} - e^{-\lambda_1 t} \right] \\ N_2 &= \frac{\lambda_1 N_0}{(\lambda_1 - \lambda_2)} \left(e^{-\lambda_2 t} - e^{-\lambda_1 t} \right) \end{split}$$

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$$\begin{array}{l} \text{at } t = 0; N_2 = 0 \\ \text{at } t \rightarrow \infty; N_2 = 0 \\ \text{Variation of } N_2 \text{ with time } t \end{array}$$

