

Gases



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- ➡ A. In gases, particles are far apart from each other.
- B. Force of attraction between the particles is negligible, and they can move freely.
- C. Gases have neither a fixed volume nor a fixed shape.
- D. The gaseous state has the highest compressibility as compared to solids and liquids.
- E. Oxygen, carbon dioxide, water vapor etc. are the examples of gases.

Gas

