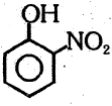
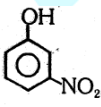
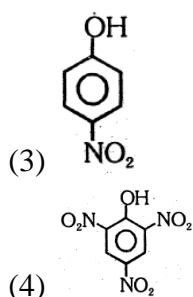


## EXERCISE-I (Conceptual Questions)

## Build Up Your Understanding

## Alcohol

- The compound A, B and C in the reaction sequence  $\text{CH}_3\text{CH}_2\text{OH} \xrightarrow{\text{PBr}_3} \text{A} \xrightarrow{\text{alc.KOH}} \text{B} \xrightarrow{\text{Br}_2} \text{C}$  are given by the set
  - (1)  $\text{C}_2\text{H}_5\text{Br}$ ,  $\text{CH}_3\text{CH}_2\text{OH}$ ,  $\text{CH}_3\text{CHBr}_2$
  - (2)  $\text{C}_2\text{H}_5\text{Br}$ ,  $\text{CH}\equiv\text{CH}$ ,  $\text{CH}_2=\text{CHBr}$
  - (3)  $\text{C}_2\text{H}_5\text{Br}$ ,  $\text{CH}_2=\text{CH}_2$ ,  $\text{CH}_2\text{Br}-\text{CH}_2\text{Br}$
  - (4)  $\text{C}_2\text{H}_5\text{Br}$ ,  $\text{CH}_3\text{CH}_2\text{OH}$ ,  $\text{BrCH}_2-\text{CH}_2\text{Br}$
- Primary amines can be converted into alkanols by reaction with
  - (1) Aqueous HCl
  - (2) Tilden's reagent
  - (3)  $\text{NaNO}_2$  and dil. HCl
  - (4) Hypochlorous acid
- Which of the following alcohols gives a red colour in Victor Meyer test
  - (1)  $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{OH}$
  - (2)  $\begin{array}{c} \text{CH}_3-\text{CH}-\text{OH} \\ | \\ \text{CH}_3 \end{array}$
  - (3)  $(\text{CH}_3)_3\text{C}-\text{OH}$
  - (4)  $\begin{array}{c} \text{CH}_3-\text{CH}-\text{CH}_2-\text{CH}_3 \\ | \\ \text{OH} \end{array}$
- Methanol on heating with salicylic acid and a few drops of cone.  $\text{H}_2\text{SO}_4$  gives the smell of
  - (1) Bitter almonds
  - (2) Oil of wintergreen
  - (3) Rotten eggs
  - (4) Mustard oil
- Which of the following does not turn orange colour of chromic acid to green
  - (1)  $1^\circ$  alcohol
  - (2)  $2^\circ$  alcohol
  - (3)  $3^\circ$  alcohol
  - (4) Allyl alcohol
- p, s and t-alcohols can be distinguished by :
  - (1) Reimer-Tiemann reaction
  - (2) Tollen's reagent
  - (3) Lucas test
  - (4) Lassaigne's test
- Consider the following reaction :
 
$$\text{C}_2\text{H}_5\text{OH} + \text{H}_2\text{SO}_4 \rightarrow \text{Product}$$
 Among the following, which one cannot be formed as a product under any conditions ?
  - (1)  $\text{C}_2\text{H}_5\text{OSO}_3\text{H}$
  - (2)  $\text{H}_2\text{C}=\text{CH}_2$
  - (3)  $\text{HC}\equiv\text{CH}$
  - (4)  $\text{CH}_3-\text{CH}_2-\text{O}-\text{CH}_3$
- Nitration of phenol with cone. nitric acid gives :-
  - (1) 
  - (2) 

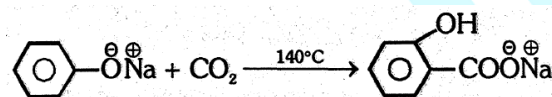
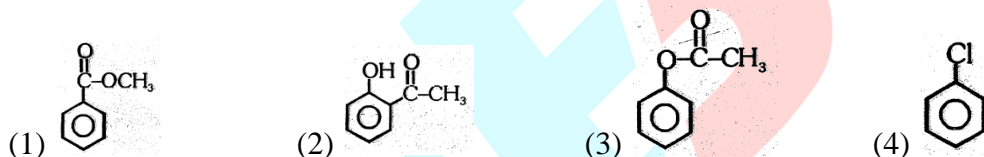
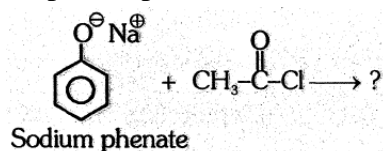


9. Deoxygenation of phenol can be achieved by distillation with :-

- (1) Raney nickel  
 (2) Lithium aluminium hydride  
 (3) Sodium borohydride  
 (4) Zinc dust

10. Which of the following compounds shows intramolecular hydrogen bonding :-

- (1) p-Nitrophenol      (2) Ethanol      (3) o-Nitrophenol      (4) Methanamine



is called:-

- (1) Schotten Bauman reaction  
 (2) Kolbe Schmidt reaction  
 (3) Reimer- Tiemann reaction  
 (4) Lederer-Manasse reaction

13. Phenol can be distinguished from ethanol by reactions with the following except :-

- (1) Iodine and alkali      (2) Ferric chloride      (3) Acetyl chloride      (4) Bromine water

14. Phenol on treatment with methyl chloride in the presence of anhydrous  $\text{AlCl}_3$  gives chiefly:-

- (1) o-cresol      (2) m-cresol      (3) anisole      (4) p-cresol

15. Phenol on heating with  $\text{NaNO}_2$  and a few drops of cone.  $\text{H}_2\text{SO}_4$  mainly gives:-

- (1) p-nitrophenol      (2) p-nitrosophenol      (3) o-nitrophenol      (4) m-nitrosophenol

16. Phenol and benzoic acid are distinguished by:-

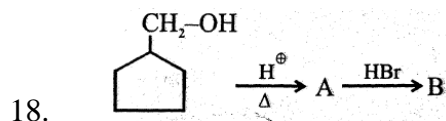
- (1) Lucas reagent      (2) Victor Meyer test      (3) Caustic soda      (4) Sodium bicarbonate

17. Phenol can be distinguished from ethanol by the reaction with

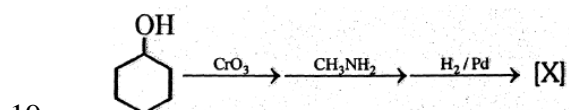
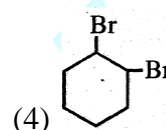
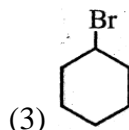
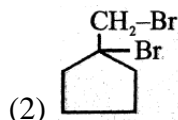
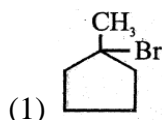
(1) Na

(2) Br<sub>2</sub>/Water(3) Neutral AlCl<sub>3</sub>

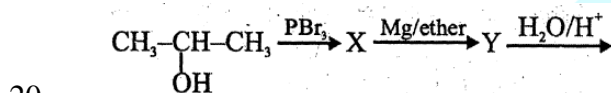
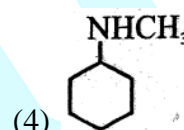
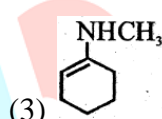
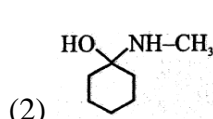
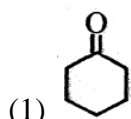
(4) 1 and 2 Both

**Ether**

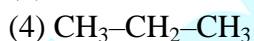
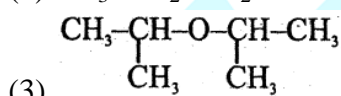
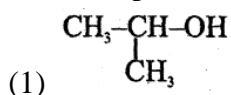
What is the structure of B :-



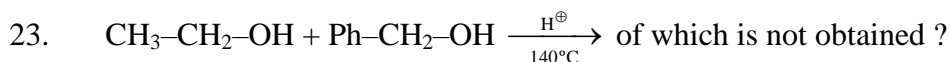
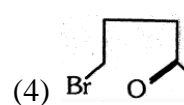
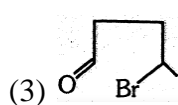
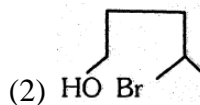
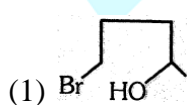
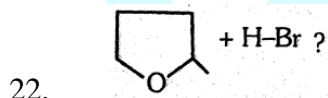
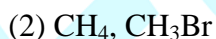
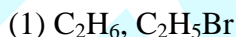
Product [X] of given reaction is :-



The final product is:



21. In the reaction sequence



$$(1) \text{C}_2\text{H}_5-\text{O}-\underset{\text{CH}_3}{\underset{|}{\text{CH}}}-\overset{\oplus}{\underset{\text{H}}{|}}{\text{O}}-\text{H} \qquad (2) \text{CH}_3-\text{CH}_2-\overset{\oplus}{\underset{\text{H}}{|}}{\text{O}}-\text{CH}_2-\text{CH}_3$$

(3)  $(\text{C}_2\text{H}_5)_2\text{O} \rightarrow \text{O}$

(4)

$$\begin{array}{c} \text{CH}_3-\text{CH}_2-\text{O}-\text{CH}_2-\text{CH}_2-\text{O}^+-\text{O}-\text{H} \\ | \\ \text{H} \end{array}$$

(1) Ethyl alcohol      (2) Ethyl iodide      (3) Methyl iodide      (4) Methanol

(1) Phenol                      (2) Benzoic acid                      (3)  $\text{CH}_3\text{COOH}$                       (4)  $\text{CH}_3\text{--O--C}_6\text{H}_5$

(1) Elemental copper  
(2) Cuprous oxide  
(3) Cupric oxide  
(4) Mixture of all of the above

- (1) A phenylhydrazone with phenylhydrazine
- (2) A sodium bisulphite adduct with sodium bisulphite .
- (3) A silver mirror with Tollen's reagent
- (4) An oxime with hydroxylamine

(1) Fehling solution    (2) Grignard reagent    (3) Schiff 's reagent    (4) Tollen's reagent

$$\text{CH}_3\text{CH}_2\text{COOH} \xrightarrow[\text{(2) Br}_2/\text{CCl}_4]{\text{(1) moist Ag}_2\text{O, } \Delta} \text{A} \xrightarrow[\text{Ether}]{\text{Mg}} \text{B} \xrightarrow{\text{CH}_3\text{CN/H}_3\text{O}^+} \text{C}$$

(1)  $\text{CH}_3\text{CH}_2-\underset{\text{OH}}{\text{CH}}-\text{CH}_3$

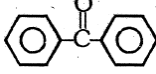
(2)  $\text{CH}_3\text{CH}_2-\text{CN}$

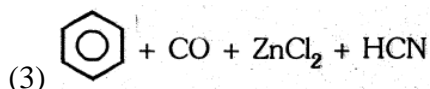
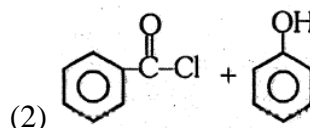
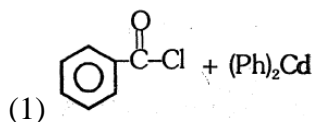
(3)  $\text{CH}_3\text{CH}_2-\underset{\text{O}}{\text{C}}-\text{CH}_3$

(4)  $\text{CH}_3-\text{CH}_2-\underset{\text{COOH}}{\text{CH}}-\text{CH}_3$

(1)  $\text{HCHO}$                       (2)  $\text{CH}_3\text{CHO}$                       (3)  $\text{CH}_3\text{COCH}_3$                       (4)  $\text{CH}_3-\text{C}=\text{N}$

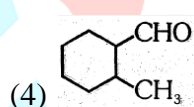
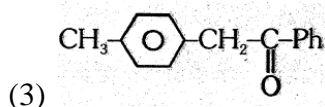
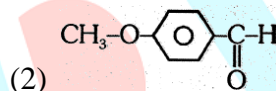
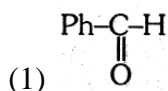
32.  $\text{Ph}-\overset{\text{Cl}}{\underset{\text{Cl}}{\text{C}}}-\text{CH}_3 + \text{aq. KOH} \rightarrow \text{A} \xrightarrow[\text{H}^+]{\text{KCN}} \text{B} ?$   
 (1) 50% d + 50% l (2) 80% d + 20% l (3) Meso compound (4) optically active

33.  can be obtained by -



(4) None of the above

34. Which does not react with  $\text{NaHSO}_3$



35. Ketones can be prepared by :-

- (1) Rosenmund reduction
- (2) Etard reaction
- (3) Cannizzaro reaction
- (4) Friedel-Craft reaction

36. Carbonyl compounds are best purified by:-

- (1) Steam distillation
- (2) Hydrolysis of sodium bisulphite adducts
- (3) fractional crystallisation
- (4) Sublimation

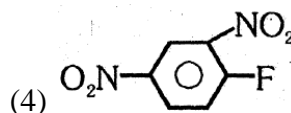
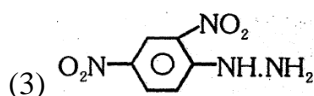
37. Carbonyl compounds readily undergo :-

- (1) Nucleophilic substitutions
- (2) Electrophilic addition reactions
- (3) Nucleophilic addition reactions
- (4) Free radical substitution reactions

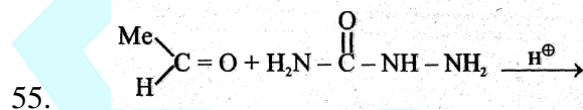
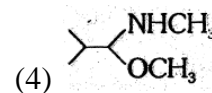
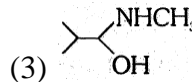
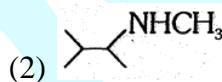
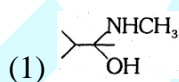
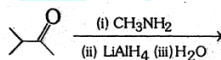
37.  $\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_3$  and  $\text{CH}_3-\overset{\text{O}}{\parallel}{\text{C}}-\text{H}$  are readily distinguished by their reaction with :-  
 (1) Iodine and alkali (2) 2,4-dinitrophenylhydrazine

- (3) Tollen's reagent (4) All the above
39. Formaldehyde and acetaldehyde are readily distinguished by reaction with :-  
 (1) A solution of 2,4-dinitrophenylhydrazine  
 (2) Fehling's solution  
 (3) Tollen's reagent  
 (4) Iodine and alkali
40. Which is the most suitable reagent for the following conversion  

$$\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{CH}_3 \rightarrow \text{CH}_3-\text{CH}=\text{CH}-\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$$
  
 (1)  $\xrightarrow[\text{(ii) H}^+]{\text{(i) Tollen's reagent}}$  (2)  $\xrightarrow[\text{(ii) H}^+]{\text{(i) Benzoyal peroxide}}$  (3)  $\xrightarrow[\text{(ii) H}^+]{\text{(i) I}_2 \text{ and NaOH}}$  (4)  $\xrightarrow{\text{KMnO}_4/\text{H}^+, \Delta}$
41. Formaldehyde reacts with cone. alkali to form :  
 (1) A resinous mass  
 (2) Formic acid  
 (3) A mixture of methanol and sodium formate  
 (4) Methanol
42. Which of the following compounds does not give aldol condensation :-  
 (1)  $\text{CH}_3\text{CHO}$  (2)  $\text{CH}_3\text{CH}_2\text{CHO}$  (3)  $\text{HCHO}$  (4)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHO}$
43. Cannizzaro reaction is given by:-  
 (1) Aldehydes containing  $\alpha$ -hydrogen atoms  
 (2) Aldehydes as well as ketones containing  $\alpha$ -hydrogen atoms  
 (3) Aldehydes not containing  $\alpha$ -hydrogen atoms  
 (4) Aldehydes containing  $\beta$ -hydrogen atoms
44. Benzaldehyde react with formaldehyde in the presence of alkali to form:-  
 (1)  $\text{CH}_3-\text{OH}$  and  $\text{C}_6\text{H}_5-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}^-\text{Na}^+$   
 (2)  $\text{C}_6\text{H}_5-\text{CH}_2-\text{OH}$  and  $\text{H}-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}^-\text{Na}^+$   
 (3)  $\text{C}_6\text{H}_5-\text{COOH}$  and  $\text{CH}_3\text{CH}_2-\text{OH}$   
 (4)  $\text{H}-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$  and  $\text{C}_6\text{H}_5-\text{CH}_2-\text{OH}$
45. Which of the following can be converted to  $\text{CH}_3-\text{CH}=\text{CH}-\text{CHO}$  :-  
 (1) Acetone (2) Acetaldehyde (3) Propanaldehyde (4) Formaldehyde
46. The product of reaction with primary amine and aldehyde is:-  
 (1)  $\text{R}-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$  (2)  $\text{R}-\text{ONO}$  (3)  $\text{R}'-\text{CH}=\text{N}-\text{R}$  (4)  $\text{R}-\text{NO}_2$
47. Brady's reagent is  
 (1)  $[\text{Cu}(\text{NH}_3)_4]\text{SO}_4$  (2)  $\text{KMnO}_4/\text{NaIO}_4$

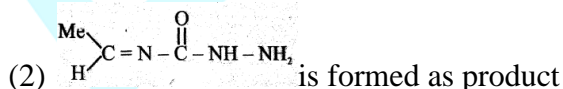


48. A compound with molecular formula  $C_3H_6O$ , not gives silver mirror with Tollen's reagent but forms oxime with hydroxyl amine. Compound will be -  
 (1)  $CH_2=CH-CH_2-OH$  (2)  $CH_3CH_2CHO$   
 (3)  $CH_2=CH-O-CH_3$  (4)  $CH_3COCH_3$
49. Aldehyde and ketone are distinguished by reagent  
 (1) Fehling solution (2)  $H_2SO_4$  (3)  $NaHSO_3$  (4)  $NH_3$
50. Carbonyl group is converted into methylene group by-  
 (1) Acidic reduction (2) Raney Ni  
 (3) Basic hydrolysis (4) Normal Hydrogenation
51. When acetaldol is treated with excess of acid then unsaturated product will be :-  
 (1) Alcohol (2) Aldehyde (3) Acid (4) Alkyl halide
52. The reagent used for the separation of acetaldehyde from acetophenone is -  
 (1)  $NaHSO_3$  (2)  $C_6H_5NHNH_2$  (3)  $NH_2OH$  (4)  $NaOH + I_2$
53. The most suitable reagent for the conversion of  
 $RCH_2OH \longrightarrow RCHO$   
 (1)  $KMnO_4$   
 (2)  $K_2Cr_2O_7$   
 (3)  $CrO_3$   
 (4) PCC (Pyridinium chloro chromate)
54. The major organic product formed from the following reaction is :-



which is incorrect :-

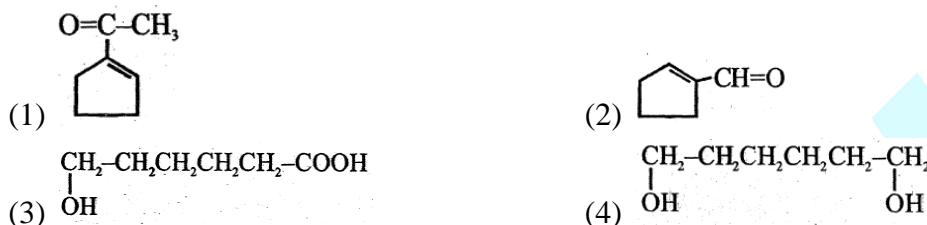
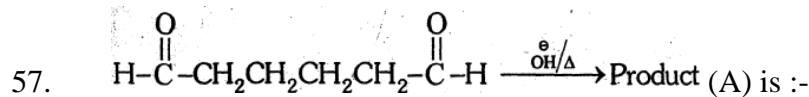
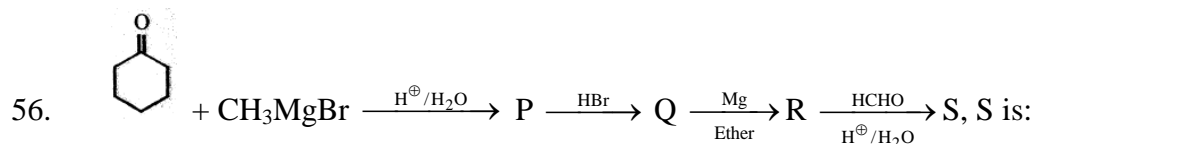
- (1) The reaction completes through addition elimination mechanism.



- (3) The reaction is an example of condensation reaction

- (4) None





58. Which of the following compounds is the product of an aldol reaction :-



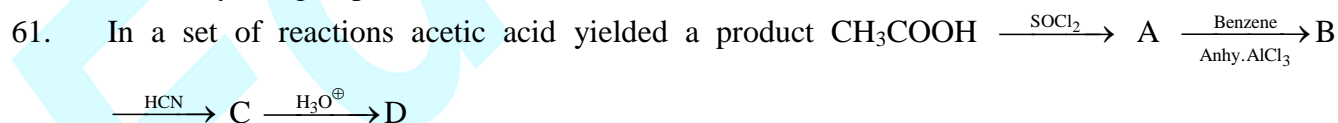
59. Which product is obtained by aldol reaction :-

- (1)  $\alpha$ -Hydroxy aldehyde or ketone
- (2)  $\beta$ -Hydroxy aldehyde or ketone
- (3)  $\alpha,\beta$ -unsaturated ester
- (4)  $\beta$ -Hydroxy acid

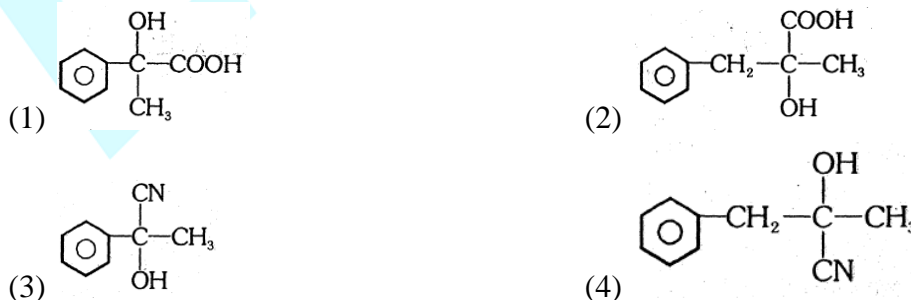
### Carboxylic acid

60. When propanoic acid is treated with aqueous sodium bicarbonate,  $\text{CO}_2$  is liberated. The C of  $\text{CO}_2$  comes from :-

- (1) methyl group
- (2) carboxylic acid group
- (3) methylene group
- (4) bicarbonate



The structure of D would be-



### Acid Derivative



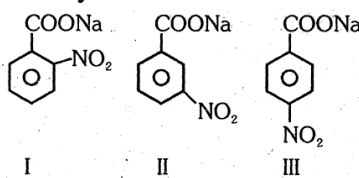
62. The compounds A and B in the reaction sequence  $B \xleftarrow{\text{Phenol}} \text{CH}_3\text{COCl} \xrightarrow{\text{CH}_3\text{COONa}} A$  are given by the set :-

- (1)  $\text{CH}_3\text{CO}-\text{O}-\text{COCH}_3$ ,  $\text{C}_6\text{H}_5\text{CH}_2\text{OH}$   
 (2)  $\text{CH}_3\text{CO}-\text{O}-\text{COCH}_3$ ,  $\text{C}_6\text{H}_5\text{OCOCH}_3$   
 (3)  $\text{CH}_3\text{COCH}_3$ ,  $\text{C}_6\text{H}_5\text{OCOCH}_3$   
 (4)  $\text{CH}_3-\text{C}(=\text{O})-\text{O}-\text{O}-\text{C}(=\text{O})-\text{CH}_3$ ,  $\text{CH}_3-\text{C}(=\text{O})-\text{O}-\text{C}_6\text{H}_5$

63.  $\text{CH}_3-\text{CH}_2-\text{COOH} \xrightarrow{\text{P/Cl}_2} A \xrightarrow[\text{KOH}]{\text{Alc.}} B$  structure of B is :-

- (1)  $\text{CH}_2=\text{CH}-\text{COOH}$  (2)  $\text{CH}_3-\text{CH}(\text{Cl})-\text{COOH}$  (3)  $\text{CH}_2(\text{Cl})-\text{CH}_2-\text{COOH}$  (4)  $\text{CH}_3-\text{CH}_2-\text{C}(=\text{O})-\text{Cl}$

64. Correct reactivity order of decarboxylation is



- (1)  $I > II > III$  (2)  $I > III > II$  (3)  $III > II > I$  (4)  $III > I > II$

65. Which is most reactive towards hydrolysis

- (1) (2)   
 (3) (4)

66. Which of the following reagents may be used to distinguish between phenol and benzoic acid ?

- (1) Victor-Mayer test (2) Neutral  $\text{FeCl}_3$   
 (3) Aqueous  $\text{NaOH}$  (4) Tollen's reagent

67. Acyl chlorides undergo :-

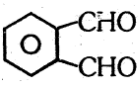
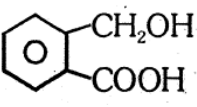
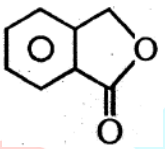
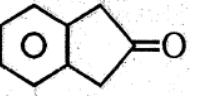
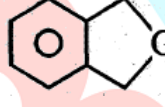
- (1) Nucleophilic addition reactions (2) Nucleophilic substitution reactions  
 (3) Electrophilic substitution reactions (4) Electrophilic addition reactions

68. The reaction of ethanol on acetic anhydride is an example of :-

- (1) Nucleophilic addition (2) Nucleophilic substitution  
 (3) Electrophilic addition (4) Free radical substitution

69. The reduction of acetamide gives :-

- (1)  $\text{CH}_3\text{CH}_2\text{NH}_2$  (2)  $(\text{CH}_3)_2\text{CHNH}_2$  (3)  $(\text{CH}_3)_3\text{CNH}_2$  (4)  $(\text{CH}_3\text{CH}_2)_2\text{NH}$

70. Which is used in preparation of aldehyde by rosenmund reduction  
 (1) Ester (2) Acid (3) Acid halide (4) Alcohol
71. Which is maximum reactive :-  
 (1)  $\text{CH}_3\text{CONH}_2$  (2)  $\text{CH}_3\text{COOCH}_3$  (3)  $\text{CH}_3\text{COCl}$  (4)  $\text{CH}_3\text{COOC}_2\text{H}_5$
72.  $\text{CH}_3\text{C(=O)NH}_2 \xrightarrow[\Delta]{\text{P}_2\text{O}_5} ?$   
 (1)  $\text{CH}_3\text{COOH}$  (2)  $\text{CH}_3\text{CN}$  (3)  $\text{CH}_3\text{CH}_3$  (4)  $\text{CH}_3\text{CHO}$
73.   $\xrightarrow{\text{NaOH}} \text{A} \xrightarrow[\Delta]{\text{H}^+} ?$
- (1)   
 (2)   
 (3)   
 (4) 

## EXERCISE-I (Conceptual Questions)

## ANSWER KEY

Que.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Ans.	3	3	1	2	3	3	3	4	4	3	3	2	3	4	2
Que.	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30
Ans.	4	2	3	4	4	3	1	4	2	2	4	2	3	2	3
Que.	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45
Ans.	4	1	1	3	4	2	3	3	4	3	3	3	3	2	2
Que.	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
Ans.	3	3	4	1	1	2	1	4	2	2	2	2	1	2	4
Que.	61	62	63	64	65	66	67	68	69	70	71	72	73		
Ans.	1	2	1	1	2	2	2	2	1	3	3	2	2		