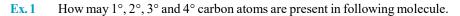
SOLVED EXAMPLES



$$CH_{3} - CH - CH_{2} - CH_{2} - CH_{3} - CH_{3$$

- **Sol.** 1° Carbon atoms = 6, 2° Carbon atoms = 2, 3° Carbon atoms = 2, 4° Carbon atom = 1
- Note: Primary, secondary, tertiary & quaternary carbon atoms in a molecule are denoted by the letters p, s, t and q respectively.
- **Ex. 2** How many 1°, 2°, 3° and 4° carbon atoms are present in following molecule.

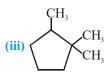
$$CH_{3} - C - CH_{2} - CH - CH_{3}$$
$$CH_{3} - C - CH_{2} - CH - CH_{3}$$
$$CH_{3} - CH_{3} - CH_{3}$$

Sol.
$$\stackrel{I^{\circ}}{\overset{I^{\circ}}{\operatorname{CH}_{3}}}_{I} - \stackrel{I^{\circ}}{\overset{I^{\circ}}{\operatorname{CH}_{2}}}_{I} - \stackrel{I^{\circ}}{\overset{I^{\circ}}{\operatorname{CH}_{2}}}_{I} - \stackrel{I^{\circ}}{\overset{I^{\circ}}{\operatorname{CH}_{3}}}_{I} - \stackrel{I^{\circ}}{\overset{I^{\circ}}{\operatorname{CH}_{3}}}_{I^{\circ}} - \stackrel{I^{\circ}}{\underset{I^{\circ}}{\operatorname{CH}_{3}}}_{I^{\circ}} - \stackrel{I^{\circ}}{\underset{I^{\circ}}{\operatorname{CH}_{3}}}$$

 1° Carbon atom = 5, 2° Carbon atom = 1, 3° Carbon atom = 1, 4° Carbon atom = 1

Ex. 3 Write the IUPAC name of following compounds. (i) $H_3C - CH_2 - CH - COOH$ I OC_2H_5

(ii) 3-Bromocyclohexane-1-sulphonic acid

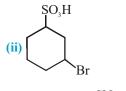


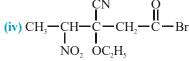
(i) 2-Ethoxybutanoic acid

(iii) 1,1,2-Trimethylcyclopentane

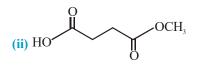
Draw the structure of following IUPAC name.

(iv) 3-Cyano-3-ethoxy-4-nitropentanoyl bromide





(ii) 3-Methoxycarbonylpropanoic acid



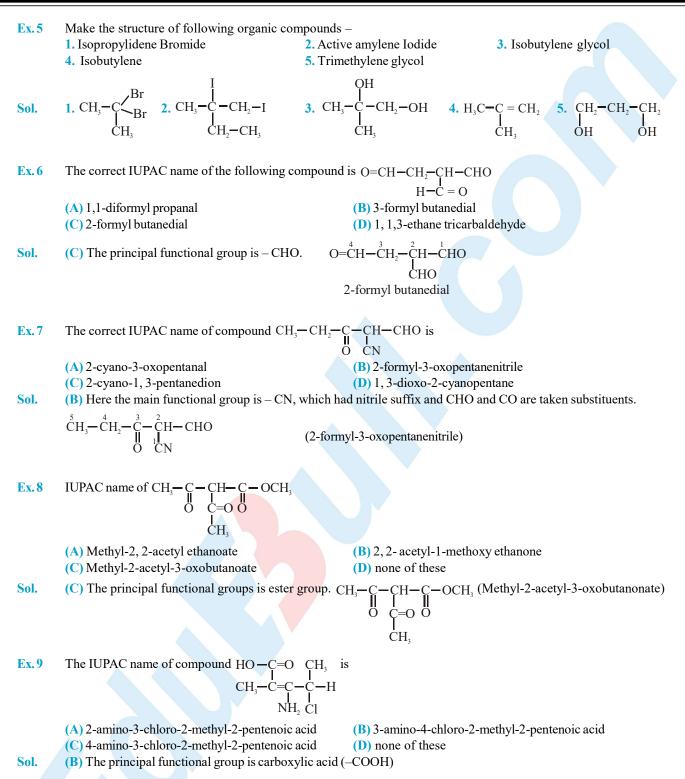
Sol. (i) 3-Ethypenta-1,4-diyne



(i)

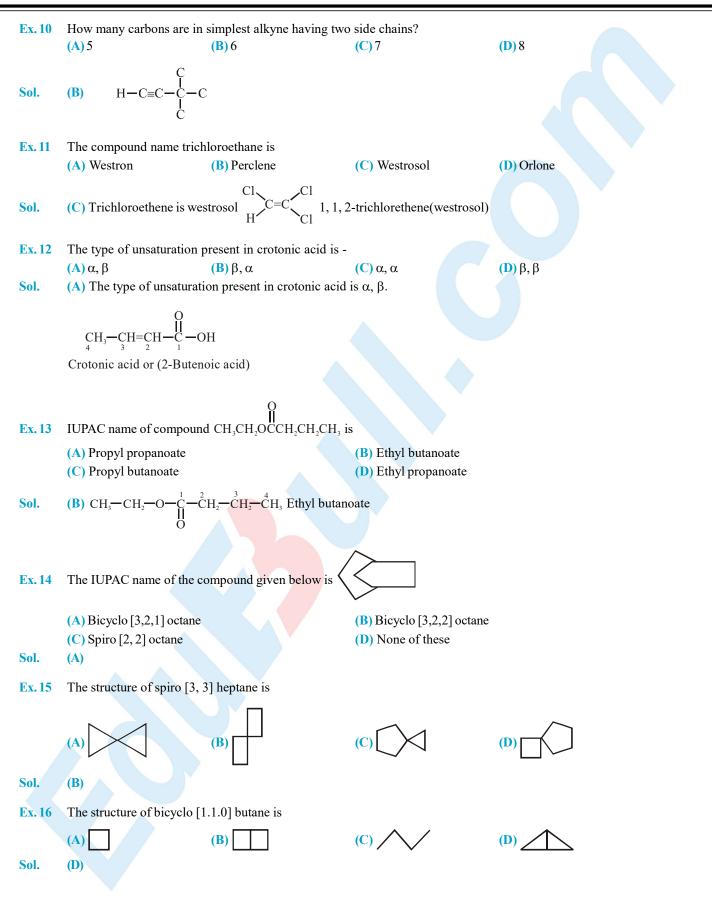
Sol.

Ex.4



$$HO - \stackrel{1}{C} = O \stackrel{5}{C}H_3$$
$$CH_3 - \stackrel{2}{C} = \stackrel{3}{C} - \stackrel{4}{C} - H$$
$$H_2 CI$$

3-amino-4-chloro-2-methyl-2-pentenoic acid



Ex	xercise # 1		[Single Correct Choice	Type Questions]
1.	The hybrid state of C-ato CH ₂ =CH-C \equiv CH	ms which are attache	d to a single bond with each oth	er in the following structure ar
	$(\mathbf{A})^{2} \operatorname{sp}^{2}, \operatorname{sp}$	(B) sp ³ , sp	(C) sp^2 , sp^2	(D) sp ² , sp ³
2.	In the compound HC \equiv C (A) sp - sp ²	$C - CH_2 - CH = CH - CI$ (B) $sp^3 - sp^3$	H_3 , the C_2 - C_3 bond is the type of : (C) sp - sp ³	(D) $sp^2 - sp^2$
3.	The number of acetynilic	bond in the structure	are : $CH \equiv C - C - CH = CH - CH = CH - CH = CH - CH = CH - CH -$	$C \equiv N$
	(A) 2	(B) 3	(C) 1	(D) 4
4.	The group of heterocyclic (A) Phenol, Furane	c compound is : (B) Furane, Thioph	ene (C) Thiophene, Phenol	(D) Furane, Aniline
5.	Which of the following is (A) Ethyl ethanoate	the first member of e (B) Methyl ethanoa	e	(D) Ethyl methanoate
6.	Which of the following co (A) Iso pentane	ompound's prefix 'iso (B) Iso Hexane	' is not correct – (C) Iso butane	(D) Iso octane
7.	A substance containing as (A) Mesityl Oxide	n equal number of pri (B) Mesitylene	imary, secondary and tertiary ca (C) Maleic acid	(D) Malonic acid
8.	How many secondary car	bon atoms does meth	nyl cyclopropane have ?	ČH ₃
	(A) Nine	(B) One	(C) Two	(D) Three
9.	The IUPAC name of the c	compound CH ₃ —CH	$= \underbrace{\mathbf{C} - \mathbf{C}\mathbf{H}_3}_{\mathbf{C}\mathbf{H}_2 - \mathbf{C}\mathbf{H}_3} \text{ is :}$	
	(A) 2-Ethyl-2-butene	(B) 3-Eth <mark>yl-2-buten</mark>	2 .	(D) 3-methyl-2-pentene
10.	IUPAC name of $CH_2 = CH$ (A) 1, 4-Hexenyne	$-CH_2-CH_2-C \equiv CH$ (B) 1-Hexen-5-yne	is : (C) 1-Hexyne-5-ene	(D) 1, 5-Hexyne
11.	$(CH_3)_3C - CH = CH_2$ has th (A) 3,3 - Dimethyl-1-bute (C) 2, 2-Dimethyl-3-buten	ne	(B) 2, 2-Dimethyl-1-bute (D) 1, 3-Dimehtyl-1-prop	
12.	What is not true about ho (A) All the members have (B) They have identical p (C) They can be represen (D) Adjacent members dif	similar chemical prop physical properties ted by a general form	ula	
13.	The homologue of pheno	l is –		
	CH ₂ OH	OH	OH L OU	OH L
	(A) (O)	(B) CH ₃	(C) OH	(D)



- 14. The IUPAC name of the following is [CH₃CH(CH₃)]₂C(CH₂CH₃)C(CH₃)C(CH₂CH₃)₂
 (A) 3,5-Diethyl-4,6-dimethyl-5-[1-methylethyl]hept-3-ene
 (B) 3, 5-Diethyl-5-isopropyl-4, 6-dimethylhept-2-ene
 (C) 3,5-Diethyl-5-propyl-4, 6-dimethylhept-3-ene
 (D) None of these
- **15.** Which of the following is a heterocyclic compound

$$\begin{array}{c} \text{HC=CH} \\ \text{(A)} \\ \text{HC=CH} \\ \text{S} \end{array} \qquad \begin{array}{c} \text{(B)} \\ \text{HC=COOH} \\ \text{HC=COOH} \end{array}$$

16. Ethyl methyl vinyl amine has the structure –

(A)
$$CH_3CH_2 - N - CH_2CH = CH_2$$

 I
 CH_3
(C) $CH_2 = CH - N - CH = CH_2$
 I
 CH_3

- 17. $CH_3-CH=CH-C=CH$, IUPAC name is: (A) Pent-2-ene-4-yne (B) Pent-4-yne-2-ene
- 18. The IUPAC name of $CH_3 C \equiv C C(CH_3)_3$ is : (A) Methyl tertirarybutyl acetylene (C) 4, 4-Dimethyl-2-pentyne
- **19.** Give the IUPAC name of

$$CH_{3}$$

$$H_{3}C - C - CH_{3}$$

$$CH_{3} - CH_{2} - CH_{2} - CH_{2} - CH_{-} - CH_{-} - CH_{2} - CH_{3}$$

$$H_{3}C - CH_{-} - CH_{-}$$

(A) 4-isopropyl-5-ter. butyl octane(C) 2-methyl-3-propyl-4-ter. butyl heptane

(B) 4-ter. butyl-5-isopropyl octane
(D) 2, 2-dimethyl-3-propyl-4-isopropyl heptane

(D) 1

(D) Pent-3-ene-1-yne

20. As per IUPAC rules, which one of the following groups, will be regarded as the principal functional group? (A) $-C \equiv C -$ (B) -OH (C) -C - (D) -C -H

(A) – (C≡C-	(B) – OH	$(\mathbf{D}) - \mathbf{C} - \mathbf{H}$

- 21. The number of C-atoms in second member of an ester is/are : (A) 2 (B) 3 (C) 4
- 22.The number of primary, secondary and tertiary carbon atom in toluene is given by the set :(A) 1, 6, 0(B) 1, 5, 1(C) 2, 5, 0(D) 1, 6, 1
- **23.** $C_3H_6Br_2$ can shows :
 - (A) Two gem dibromide(C) Two tert. dibromo alkane

(B) Three vic dibromide(D) Two sec. dibromo alkane

(B) $CH_3CH_2 - N - CH = CH_2$ I CH_3

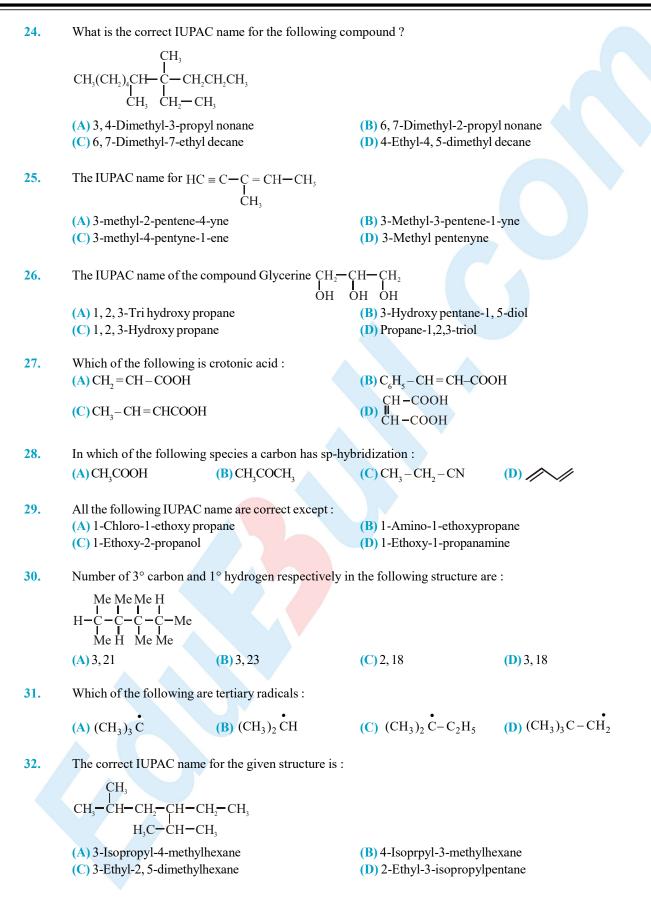
 $(D) CH_3 - N - CH = CH_2$ $I \\ CH_3$

(C) Pent-1-yne-3-ene

(B) t-Butyl propyne

(D) 1, 3, 3, 3 - Tetramethyl ethyne







	Et Me	
33.	The IUPAC name of is :	
	 (A) 2, 3-Dimethyl hexane (C) 3-Ethyl-2-methyl pentane 	(B) 2-Ethyl-4-methyl pentane(D) 2, 4-Dimethyl hexane
		Ph
34.	The IUPAC name of the compound is CH ₃ —CH—C	CH—NH ₂
	 (A) 1-Amino-1-phenyl-2-methyl propane (C) 2-Methyl-1-amino-1-phenyl propane 	(B) 2-Methyl-1-phenyl propane-1-amine(D) 2-Chloro-2-Methylpropane
35.	The IUPAC name of the compound Br (Cl) CH.CF ₃ i (A) haloethane (C) 2-bromo-2-chloro-1, 1, 1-triflouroethane	 (B) 1, 1, 1-triflouro-2-bromo-2-chloroethane (D) 1-bromo-1-chloro-2, 2, 2-triflouro ethane
36.	IUPAC name of compound is $CH_3 - CH - CH_2 - C$	$H(OH) - CH_3$
	IUPAC name of compound is CH_3 — CH — CH_2 — C CH_2CH_3	
	(A) 4-methyl-3-hexanol (B) heptanol	(C) 4-methyl-2-hexanol (D) none of these
37.	The IUPAC name of tert-butyl chloride is : (A) 4-Chlorobutane (C) 3-Ethyl-2-methyl pentane	(B) 2-Ethyl-2-methyl pentane(D) 2-Chloro-2-Methyl propane
	/	
38.	The IUPAC name of is	
	(A) 4-ethyl-3-methyl hexane(C) 3-methyl-4-ethyl hexane	(B) 3-ethyl-4-methyl hexane(D) None of these
39.	The correct nomenclature (IUPAC) for the following	g alcohol is :
	CH ₃ CH ₂ CH ₂ CH ₃	
	CH ₃ CH ₂ OH	
	(A) 2-Ethyl-2-butanon(C) 3-Ethyl pentan-3-ol	(B) 1-Ethyl-1-methyl-pentanol-1(D) diethyl ethanol
40.	The IUPAC name of is :	
	 (A) 1, 1-diethyl-2, 2-dimethyl pentane (C) 5, 5-diethyl-4, 4-dimethylpentane 	(B) 4, 4-dimethyl-5, 5-diethylpentane(D) 3-ethyl-4, 4-dimethylheptane
41		
41.	Underline carbon is sp ³ hybridised in : (A) $CH_3 - CH = CH_2$ (C) CH_3CONH_2	(B) $CH_3CH_2 - NH_2$ (D) CH_3CHCN
	3 2	··· 3
42.	The IUPAC name of is -	
	(A) 2-ethyl-3-methyl-1-penten-4-yne (C) 4-ethyl-3-methyl-1-pentyn-4-ene	(B) 2-ethyl-3methyl-4-pentyn-1-ene(D) 4-ethyl-3-ethyl-4penten-1-yne



43. The correct IUPAC name of :

(A) 3-methyl pentanoyl chloride(C) 1-chloro-3-ethyl butanone

(B) 3-methyl butanoyl chloride(D) 1-chloro-3-methyl pentanone

44. The correct IUPAC name of CH_3 — CH_2 —C—COOH

(A) 2-methyl butanoic acid(C) 2-carboxy-1- butene

45. IUPAC name will be
$$CH_2$$
— CH — CH_2
I I I
CN CN CN

(A) 1, 2, 3-Tricyano propane (C) 1, 2, 3-Cyano propane (B) 2-ethyl-2-propenoic acid(D) None of these

(B) Propane-1,2,3-trinitrile(D) Propane-1, 2, 3-tricarbonitrile

46. The IUPAC name of compound

(A) 1, 2, 3-Tricarboxypropan-2-ol
(B) 2-Hydroxy propane-1, 2, 3-tricarboxylic acid
(C) 3-Hydroxy-3-carboxypentane-1, 5-dioic acid
(D) None

CH.-COOH

47. The IUPAC name of the structure is :

H₂N-CH-CH-CHO I I HOOC COOH

(A) 3-Amino-2-formyl butane-1, 4-dioic acid
(C) 2-Amino-3-formyl butane-1, 4-dioic acid

(B) 3-Amino-2, 3-dicarboxy propanal(D) 1-Amino-2-formyl succinic acid

48. Which of the following compound is wrongly named ?

Column I (A) $CH_3CH_2CH_2CHCOOH$ Cl (B) $CH_3C \equiv CCHCOOH$ CH₃ (C) $CH_3CH_2CH = CHCOCH_3$ (D) $CH_3 - CHCH_2CH_2CHO$ CH₃

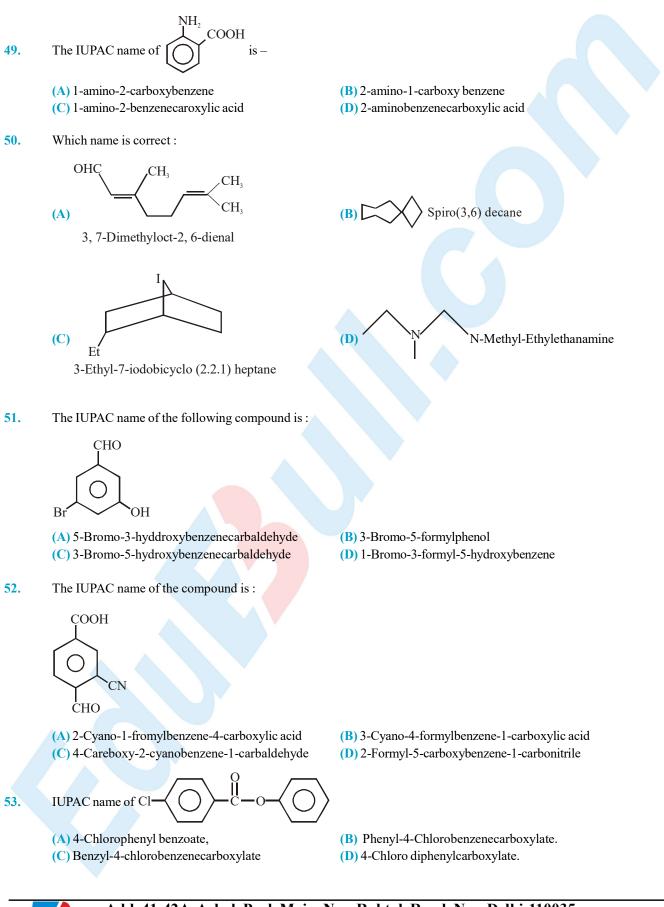
Column II

2-Chloro pentanoic acid

2-Methyl hex-3-enoic acid

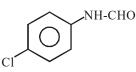
Hex -3- en - 2- one 4-Methyl pentanal







54. The correct IUPAC name of the compound.



(A) N-Formyl-4-chlorobenzenamine

(B) N-Formyl-4-chloroaniline

(C) N-(4-chlolrophenyl)methanamide

(D) N-(Parachlorophenyl)–N-Formylaniline

IUPAC name of the compound

55.

(A) 2-Chlorocarbonyl ethylbenzenecarboxylate

COOC₂H₅

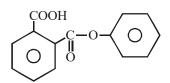
COCI

is

(B) 2-Carboxyethylbenzoyl chloride

(C) Ethyl 2-(chlorocarbonyl)benzenecarboxylate

- (D) Ethyl 1-(chlorocarbonyl)benzenecarboxylate
- **56.** The correct IUPAC name of the compound



- (A) 2-Phenoxycarbonylbenzenecarboxylic acid
- (B) Phenyl-2-carboxybenzenecarboxylate
- (C) 2-Benzoyloxybenzenecarboxylic acid
- (D) 2-Bezyloxycarbonylbenzenecarboxylic acid



Exercise # 2 Part # I [Multiple Correct Choice Type Questions]

Which of the following statement is /are wrong?

1.

- (A) $C_n H_{2n}$ is the general formula of alkanes
- (B) In homologous series, all members have the same physical properties
- (C) IUPAC means International Union of Physics and Chemistry
- (D) Butane contains two 1° C atoms and 2°C atom
- 2. Which of the following statement is/are correct?
 - (A) Homologous series can be represented by a general formula
 - (B) The chemical properties of an organic compound depend on the functional group
 - (C) Group obtained by the removal of one H atom from the alkane are called alkyl groups
 - (D) Alkynes consist of one double-bond in their molecules
- 3. Which of the following statement is/are correct ?
 - (A) Methane was named as fire damp as it formes explosive mixture with air
 - (B) Primary suffixes are added to the root word to show saturation or unsaturation in a C atom
 - (C) The IUPAC name of the valeric acid is pentanoic acid
 - (D) The common name of hexanoic acid is caproic acid
- 4. Which of the following statement is /are correct?
 - (A) The IUPAC name of amyl alcohol is pentanol
 - (B) The IUPAC name of isoamyl alcohol is 3-methyl butanol
 - (C) Wood spirit is methanol
 - (D) Methyl alcohol is also called carbinol
- 5. Which of the following statement is/are correct?
 - (A) The trivial names of organic compounds are called common names
 - (B) The systematic names of organic compound are obtained from the IUPAC system
 - (C) The systematic name of alkanes are based on the number of C atom in the longest continues chain of C atoms

(D) The maximum number of functional groups must be included in the C atom chain selected even if it does not satisfy the longest chain rule

6. Which of the following statement is/are wrong?

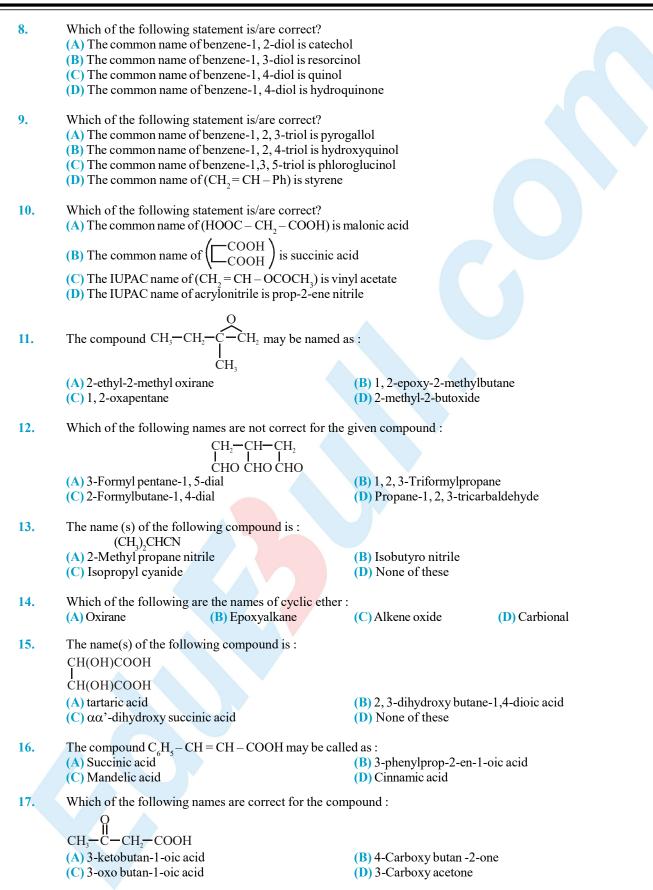
(A) Acetic acid is the systematic of vinegar

- (B) Me-C-OH is an unsaturated compound
- (C) Prefixes like n-, iso, sec-, tert, neo- etc. are used in IUPAC system.
- (D) The systematic names of acids are formed by dropping –e of the name of parent alkane and adding –oic acid.
- 7. Which of the following statement is /are correct?

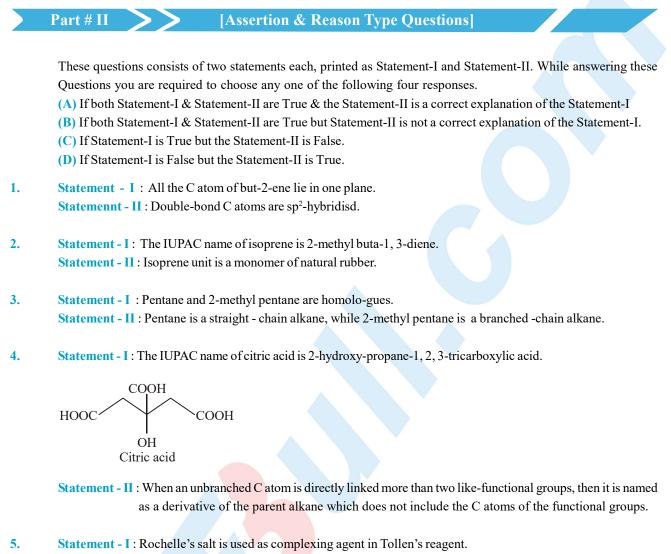
(A) R - C - O - C - R is an unsaturated compound

- (B) Neohydrocarbons contain a 3° C atom
- (C) The IUPAC name of isopropyl alcohol is propan-2-ol
- (D) The IUPAC name of (CH₃CN) is ethanenitrile

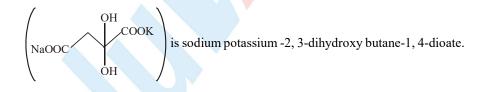








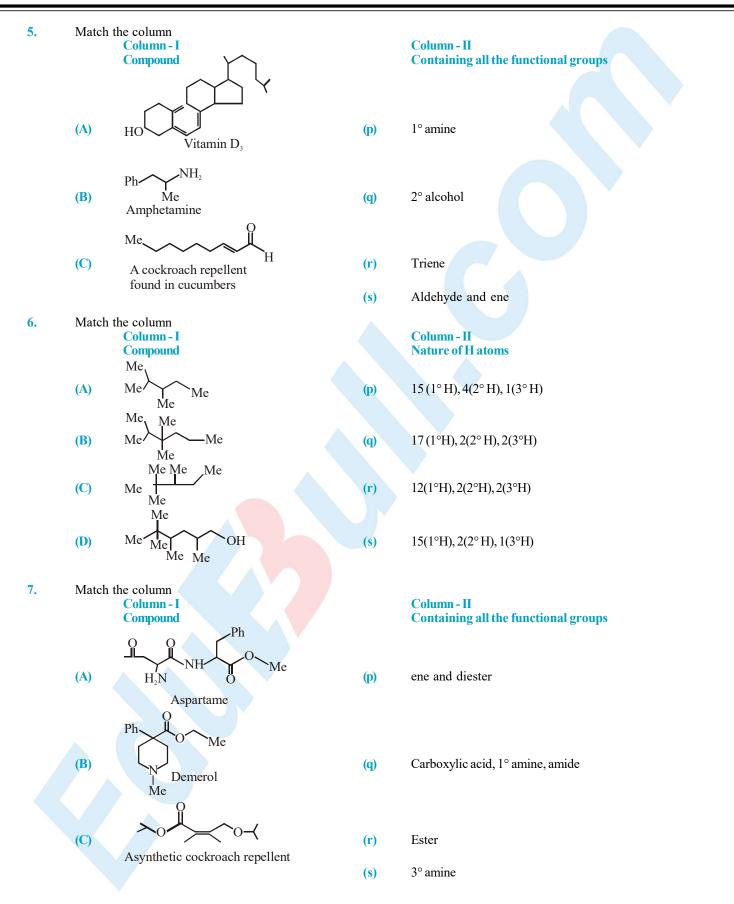
Statement -II : Sodium potassium salt of tartaric acid is known as Rochelle's salt. The IUPAC name of Rochelle's salt



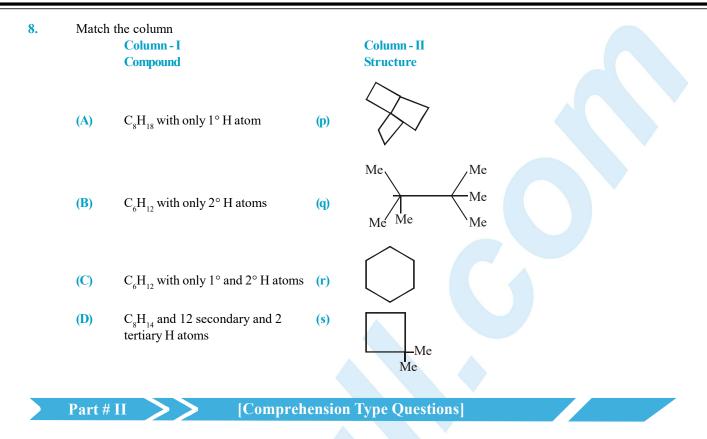


zer	cise # 3 Part #		[Matrix Match Type Questions]
Mate	h the following the compounds	of column I wit	h column II.
	Column - I		Column - II
(A)	$C_{n}H_{2n+2}$	(p)	Alkynes
(B)	$C_{n}H_{2n}$ $C_{n}H_{2n-2}$	(q)	Alkenes
(C)	C_nH_{2n-2}	(r)	Cyclohexane
(D)	$C_{6}H_{12}$	(\$)	Paraffins or alkanes
Mate	th the following the compounds of	of column - I w	
	Column - I		Column - II
(A)	Wood spirit	(p)	2-Butyne
(B)	Acetone	(q)	Trichloromethane
(C)	Dimethyl acetylene	(r)	Methanol
(D)	Chloroform	(\$)	Propanone
Mate	h column I with column II and s	elect the correc	ct answer from the given codes :
	Column - I		Column - II
	(Compounds)		(number of carbons in the bridges)
(A)		(p)	[3.2.1]
(B)	\bigwedge	(q)	[4.3.0]
	\checkmark		
(C)		(r)	[4.4.0]
(D)		(s)	[3.2.0]
(D)			[5.2.0]
Mate	the column Column - I		Column - II
	Structure		Common name
			~ JIIIII IIIIII
(A)	Me – C – Me	(p)	Caproic acid
	$\frac{4}{2}$		Carbinal
(B)	Me ⁵ 3 COOH	(q)	Carbinol
(C)	Me ⁶ <u>4</u> <u>2</u> COOH	(r)	Acetone
(D)	CH ₃ OH	(s)	Valeric acid
			С ООН
(E)	PhOH	(t)	< _{соон}
(F)	Malonic acid	(u)	Carbolic acid









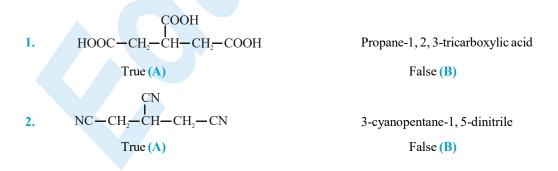
Comprehension #1

If the organic compound contains more than two similar terminal groups and all of them are directly attached to the principal chain, then none of them forms a part of the principal chain. Special suffixes are used to name these :-

Functional group	Suffix
CONH ₂	Carboxamide
—CN	Carbonitrile
—СНО	Carbaldehyde
—СООН	Carboxylic acid

Carbon atoms of these terminal groups are not counted in the principal chain. If any one of these terminal groups is not directly attached to the parent chain and forms the part of side chain, then the longest chain is selected containing two such similar groups at its two ends. The groups present in the side chain are treated as substituents and are indicated by suitable prefixes.

Indicate whether the following IUPAC names are true (A) or false (B)





	CH_2COOH	
3.	$HOOC-CH_2-CH-CH_2-COOH$	3-(carboxymethyl)-1, 5-dioic acid
	True (A)	False (B)
	CH ₂ CHO	
4.	OHC-CH ₂ -CH-CH ₂ -CHO	3-(formylmethyl)pentane-1, 5-dial
	True (A)	False (B)
	CONH ₂	
5.	$H_2NOC - CH_2 - CH - CH_2 - CONH_2$	Propane-1, 2, 3-tricarboxamide
	True (A)	False (B)

Comprehension #2

In addition to the standard ring systems (such as cyclohexane), cyclic compounds can also be bicyclic, tricyclic, etc. or they can be spirocyclic, bicyclic or bridge head carbons. The point of attachment of two rings are called bridge head atoms.

The formal names of bicyclic and related ring systems are based on

(A) Total number of atoms in the molecule.

(B) The number of atoms in each bridge connecting the bridge head atoms. These numbers are written in square bracket in decreasing order.

Spirocyclic compounds have two fused rings, but only bridge head atom. Spirocyclic compounds are named like bicyclic compounds, but have the prefix spirocyclo. Answer the following question :

 What is the IUPAC name of the above compound ?

 (A) cyclo [1.2.2] heptane

 (C) Bicyclo [2.2.1] heptane

(B) Bicyclo [1.2.2] heptane(D) cyclo [2.2.1] heptane

2.

The number of atoms in each bridge are :

(B)[3.1.0]

(A) [3.2.1]

(C) [1.3.0]

(D) [2.1.0]

3. Select the correct statement about the following compounds :

(A) It is a tricyclic compound(C) It is spiro compound

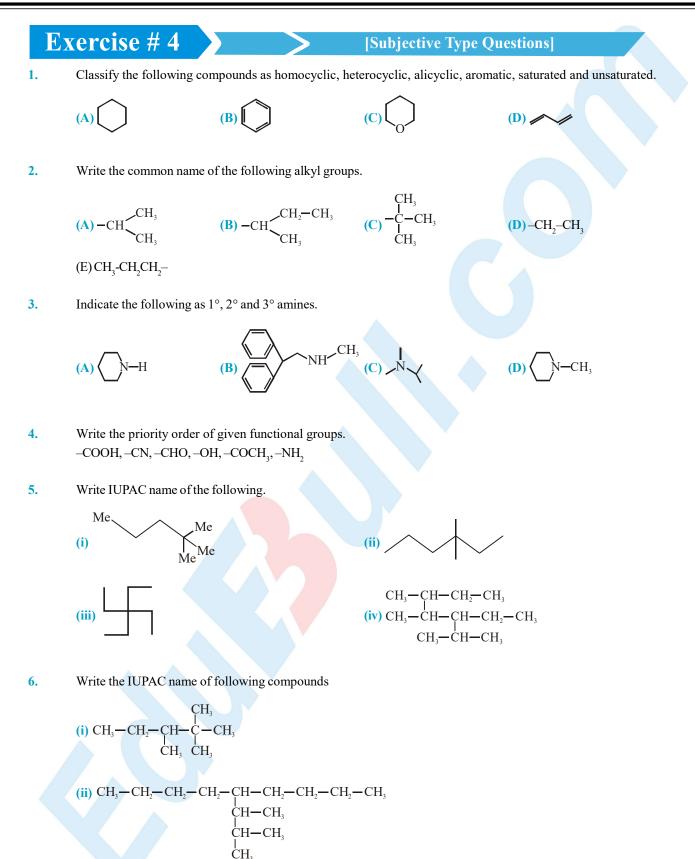
(B) It is bicyclo compound(D) Its IUPAC name is bicyclo [2.2.2] hexane

4.

Which of the following is the correct structure of bicyclo [1.1.0] butane?

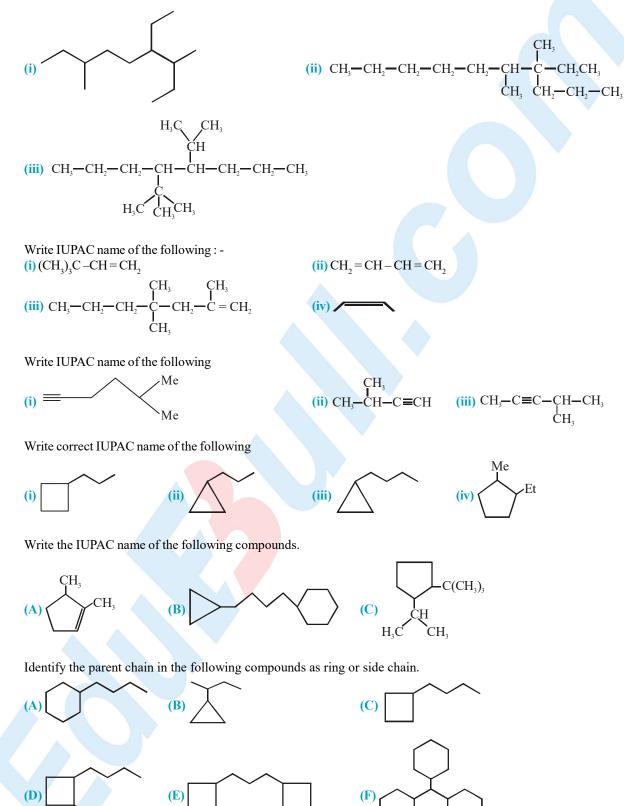








7. Write the correct IUPAC name of the following compounds.





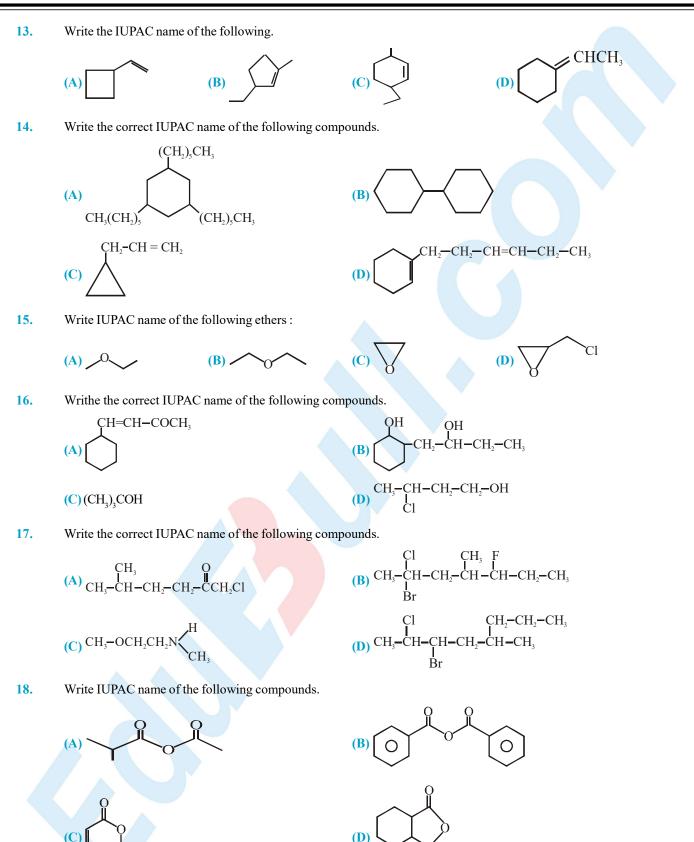
8.

9.

10.

11.

12.





19. Write IUPAC name

0

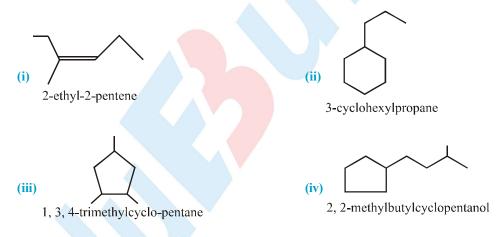
20.

Write the correct IUPAC name of the following compounds.

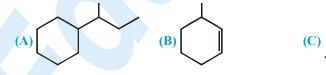
(A)
$$CH_3 - CH - CH_2 - COOH$$

 I_{CH_3}
(B) $CH_3 - CH_2 - CH - CH_2 - CH_2 - CHO$
 CH_3
(C) $CH_3 - CH_2 - CH_2 - CH_2 - CH$
(D) $CH_3 - CH_2 - CH_2 - CH_2 - CH_2 - CH_2$
(D) $CH_3 - CH_2 -$

- 21. A certain substances contains only carbon and hydrogen and has a molecular weight of 70. Photochemical chlorination gave only one monochloride. Write the structure and IUPAC name of the hydrocarbon and its monochloride.
- 22. A hydrocarbon of molecular weight 72 g mol⁻¹ has a 2-methyl group. What is the IUPAC name? Also drawn its bond-line structure ?
- Write the structure and give IUPAC systematic name of an alkane or cycloalkane with the formula :
 (A) C₈H₁₈ that has only primary hydrogen atoms
 (B) C₆H₁₂ that has only secondary hydrogen atoms.
- 24. What is wrong with the names given for these compounds provide the correct name for each :



25. Write the IUPAC name for each the following structures :



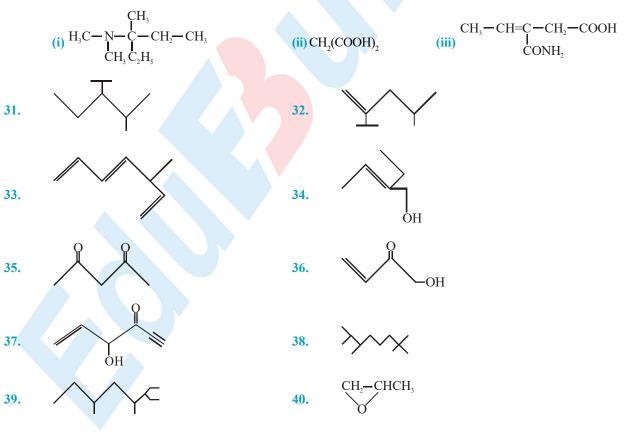
26. Write down the correct priority for citation as principal groups :

, --COOH, --CHO, --OH, --Br, -.

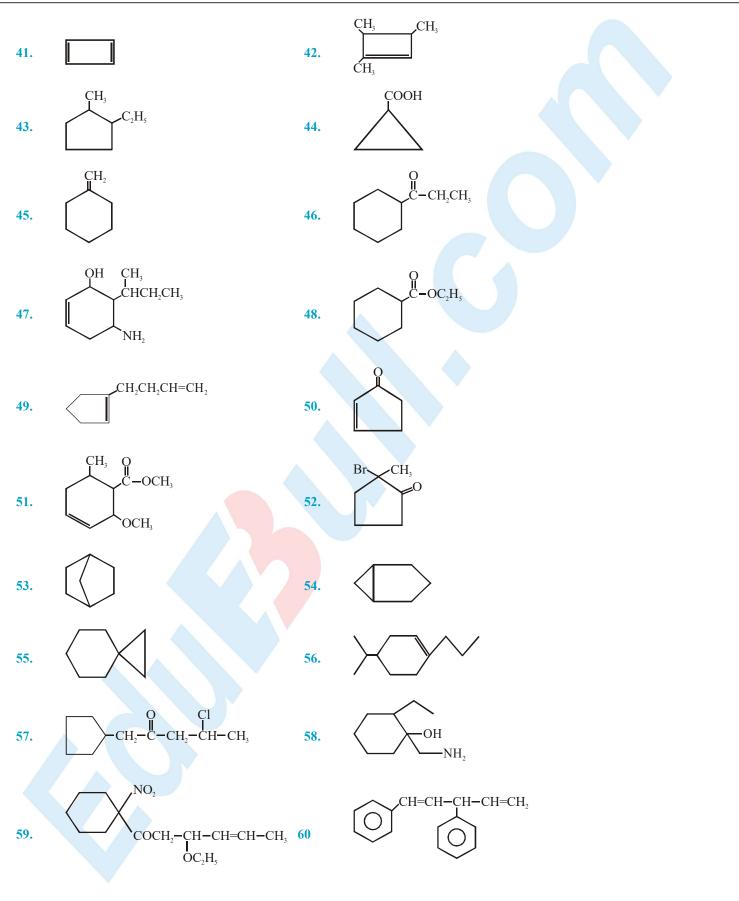


27. Write down the correct IUPAC name of the following compounds :

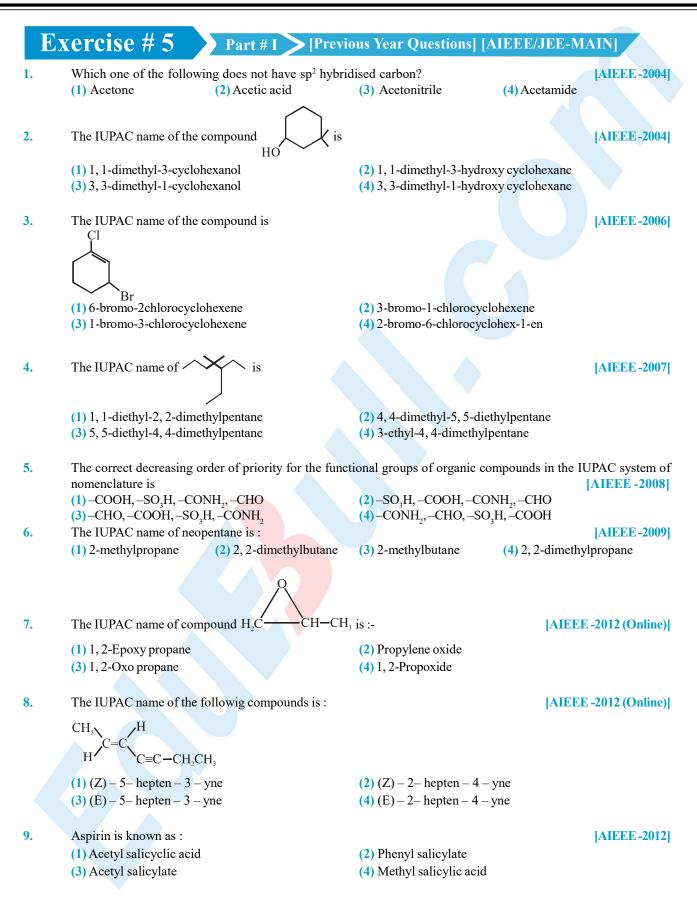
- 28. Write down the structure of the given compounds :
 (i) Bircylo [4.3.1] decane
 (ii) 1-(3'-methylcyclopentyl) benzene
 (iii) 4-ethyl-2-methyl-1-propylcyclohexane
- **29.** Answer the following :
 - (i) What would be the molecular formula for a straight chain hydrocarbon having 8 carbon atoms with (A) All C–C single bond,
 - (B) Three C–C double bond,
 - (C) one C–C triple bond and one C–C double bond.
 - (ii) What is the minimum number of carbon atoms in
 - (A) a branched alkane.
 - (B) cyclo-alkane
- **30.** Give the IUPAC names of the following compounds :



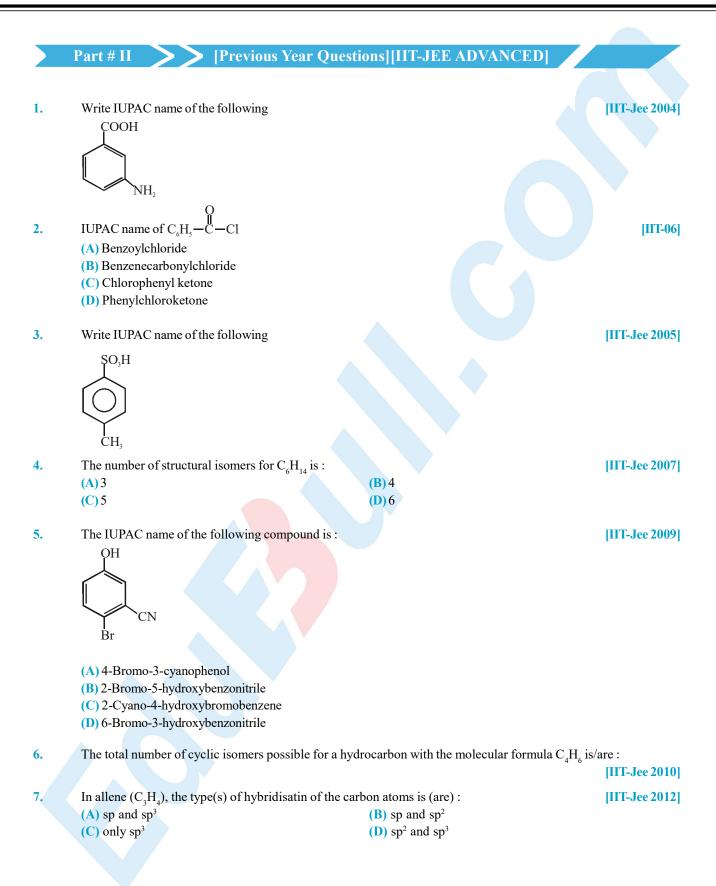












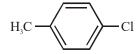


8. The carboxyl function group (-COOH) is present in :
 (A) picric acid
 (B) barbituric acid
 (C) ascorbic acid
 (D) aspirin

[IIT-Jee 2012]

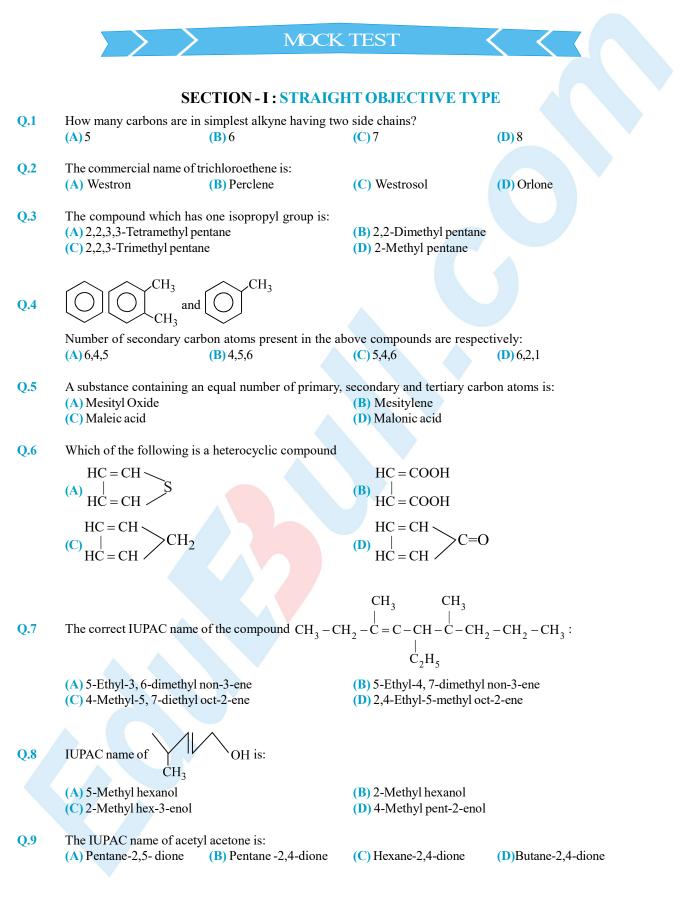
[IIT-Jee 2017]

9. The IUPAC name(s) of the following compound is(are)

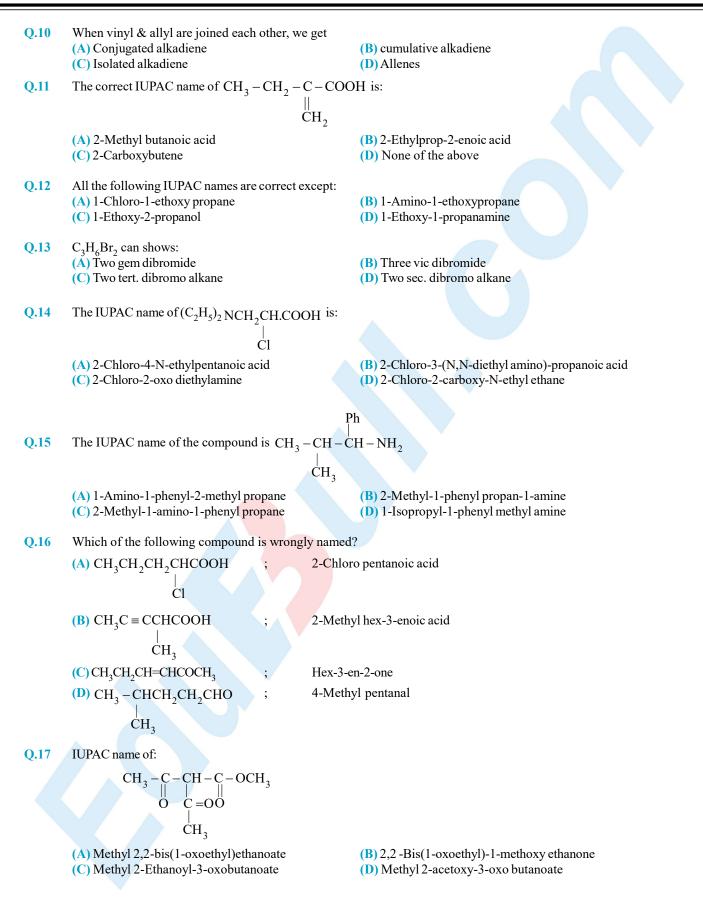


- (A) 4-methylchlorobenzene
- (B) 4-chlorotoluene
- (C) 1-chloro-4-methylbenzene
- (D) 1-methyl-4-chlorobenzene

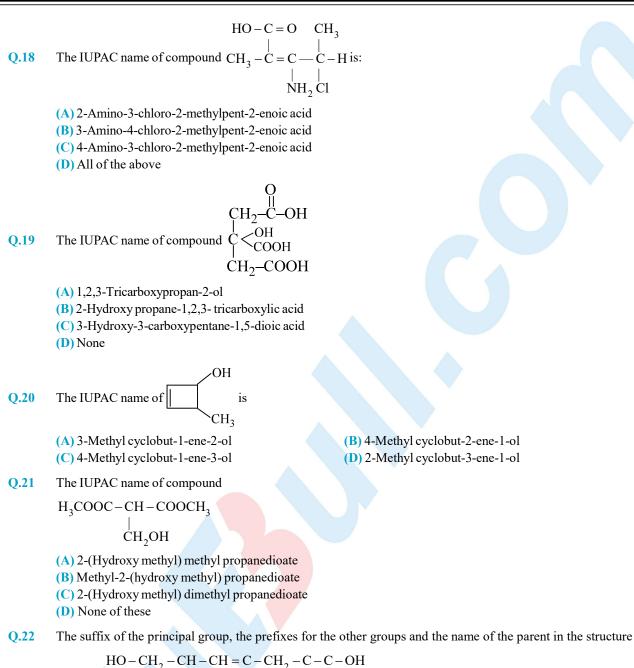












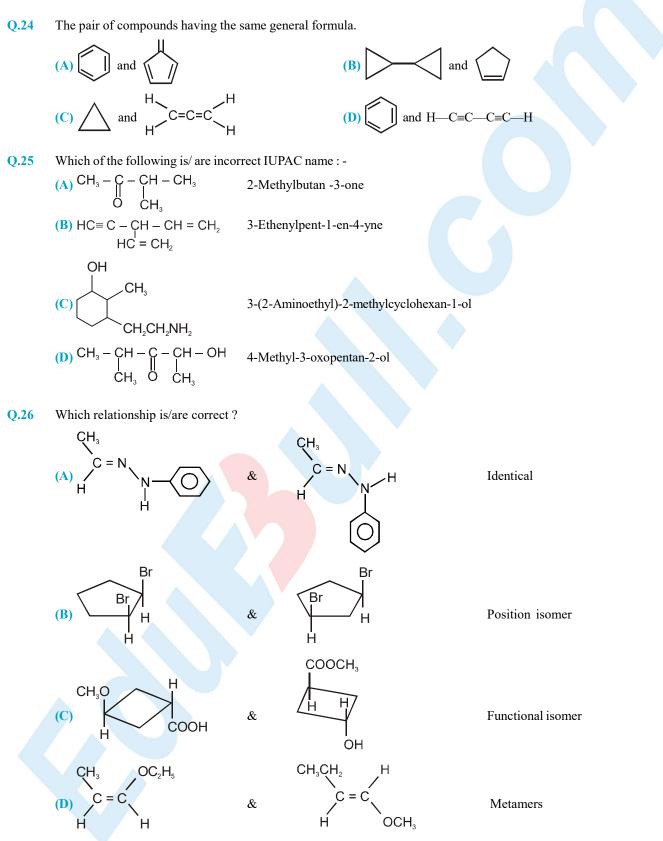
 $\begin{array}{c|c} \mathrm{HO}-\mathrm{CH}_{2}-\mathrm{CH}-\mathrm{CH}=\mathrm{C}-\mathrm{CH}_{2}-\mathrm{C}-\mathrm{C}-\mathrm{OH}\\ & & | & | & ||\\ \mathrm{CH}_{3} & \mathrm{Cl} & \mathrm{O} & \mathrm{O} \end{array}$

(A) -oic acid, chloro, hydroxy, oxo, methyl, hept-4-ene
(B) -oic acid, chloro, hydroxy, methyl, oxo, hept-4-ene

- (C) -one, carboxy, chloro. methyl, hydroxy, hept-4-ene
- (D) -one, carboxy, chloro, methyl, hydroxy, hept-4-ene
- Q.23 The IUPAC name of β -ethoxy- α -hydroxy propionic acid (trivial name) is:
 - (A) 1,2-Dihydroxy-1-oxo-3-ethoxy propane
 - (B) 1-Carboxy-2-ethoxy ethanol
 - (C) 3-Ethoxy-2-hydroxy propanoic acid
 - (D) All above



SECTION - II : MULTIPLE CORRECT ANSWER TYPE





SECTION - III : ASSERTION AND REASON TYPE

Each question has 5 choices (A), (B), (C), (D) and (E) out of which ONLY ONE is correct.

(A) Statement-1 is true, Statement-2 is true and Statement-2 is correct explanation for Statement-1.

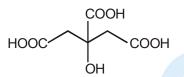
(B) Statement-1 is true, Statement-2 is true and Statement-2 is not correct explanation for Statement-1.

(C) Statement-1 is true, Statement-2 is false.

(D) Statement-1 is false, Statement-2 is true.

(E) Both Statements are false.

Q.27 Statement-1: The IUPAC name of citric acid is 2-hydroxy propane 1,2,3,- tricarboxylic acid



Statement-2: When an unbranched carbon chain is directly linked to more than two like functional groups, then it is named as derivative of parent alkane which does not include the C-atoms of the functional groups.

- Q.28 Statement-1 : The IUPAC name for the compound, OHC–CH₂–CH₂–COOH is butane -3-formyl-1-oic acid Statement-2 : COOH is considered as substituent group while CHO is considered as the principal functional group.
- Q.29 Statement-1 : The IUPAC name for the compound C₆H₅ COOCH₂CH₂COOH is 3-benzoyloxypropanoic acid. Statement-2 : C₆H₅CH₂O is called benzoyloxy group
- Q.30 Statement-1 : Pentane and 2-methyl pentane are homologues Statement-2 : Pentane is straight chain alkane, while 2-methyl pentane is a branched chain alkane.
- Q.31 Statement-1 : Butane and 2-methyl butane are chain isomers Statement-2 : Butane is a straight chain alkane while 2-methyl butane is a branched chain alkane.
- Q.32 Statement-1 : Neopentane is chain isomer of n-pentane. Statement-2 : Molecular formula of neopentane and n-pentane is C_5H_{12} .

SECTION - IV : COMPREHENSION TYPE

Comprehension

A saturated hydrocarbon (P) has six membered ring. Three alkyl groups attached to the ring alternate to each other.

(i) First group has only two carbon atoms.

(ii) Second group has four carbon atoms and its all hydrogen atoms are chemically same.

(iii) Third group has total five carbon atoms. Its main chain contains three carbon atoms with ethyl as a substituent.

- Q.33How many 3° hydrogen atoms are present in the hydrocarbon (P) ?
(A) 2(B) 3(C) 4(D) 5Q.34How many 2° carbon present in the compound (P).
(A) 10(B) 12(C) 6(D) 8
- Q.35 IUPAC name of hydrocarbon (P) is

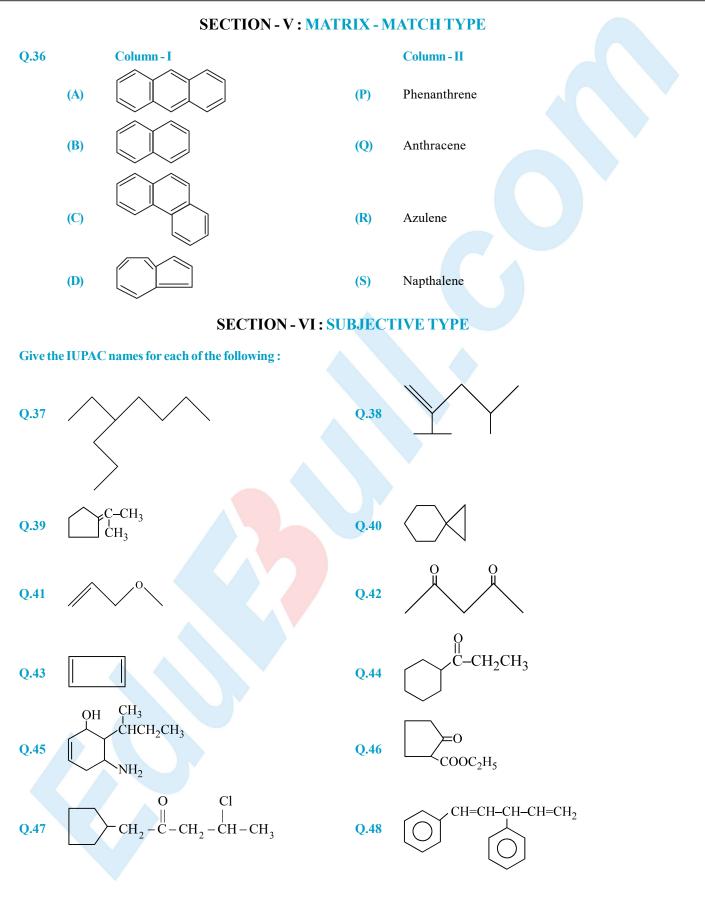
(A) 1-(1-Ethylpropyl)-3-ethyl-5-(1,1-dimethylethyl)cyclohexane

(B) 1–Ethyl–3–(1–ethylpropyl)–5–(1,1–dimethylethyl)cyclohexane.

(C) 1–(1,1–Dimethylethyl)–3–ethyl–5–(1–ethylpropyl)cyclohexane

(D) 1-(1,1-Dimethylethyl)-3-ethyl-5-(2-ethylpropyl)cyclohexane







ANSWER KEY

EXERCISE - 1

 1. A
 2. C
 3. C
 4. B
 5. C
 6. D
 7. B
 8. C
 9. D
 10. B
 11. A
 12. B
 13. B

 14. A
 15. A
 16. B
 17. D
 18. C
 19. B
 20. D
 21. B
 22. B
 23. A
 24. D
 25. B
 26. D

 27. C
 28. C
 29. B
 30. A
 31. C
 32. C
 33. D
 34. B
 35. C
 36. C
 37. D
 38. B
 39. C

 40. D
 41. B
 42. A
 43. A
 44. B
 45. D
 46. B
 47. C
 48. B
 49. D
 50. B
 51. C
 52. B

 53. B
 54. C
 55. C
 56. A
 56. A
 56. A
 56. A
 56. A

EXERCISE - 2 : PART # I

 1. A, B, C
 2. A, B, C
 3. A, B, C, D
 4. A, B, C, D
 5. A, B, C, D
 6. A, B, C
 7. C, D

 8. A, B, C, D
 9. A, B, C, D
 10. A, B, D
 11. A, B
 12. B, C
 13. A, B, C
 14. A, B, C

 15. A, B, C
 16. B, D
 17. A, C
 11. A, B
 12. B, C
 13. A, B, C
 14. A, B, C

PART # II

1. A 2. B 3. B 4. A 5. A

EXERCISE - 3 : PART # I

- 1. $A \rightarrow (s), B \rightarrow (q, r), C \rightarrow (p), D \rightarrow (q, r)$
- 2. $A \rightarrow (r), B \rightarrow (s), C \rightarrow (p), D \rightarrow (q)$
- 3. $A \rightarrow (q), B \rightarrow (r), C \rightarrow (s), D \rightarrow (p)$
- 4. $A \rightarrow (r), B \rightarrow (s), C \rightarrow (p), D \rightarrow (q), E \rightarrow (u), F \rightarrow (t)$
- 5. $A \rightarrow (r, q), B \rightarrow (p), C \rightarrow (s)$
- 6. $A \rightarrow (r), B \rightarrow (p), C \rightarrow (s), D \rightarrow (q)$
- 7. $A \rightarrow (q, r), B \rightarrow (r, s), C \rightarrow (p)$
- 8. $A \rightarrow (q), B \rightarrow (r), C \rightarrow (s), D \rightarrow (p)$

PART # II

Comprehension #1:	1.	Т	2.	F	3.	F	4.	Т	5.	Т
Comprehension #2:	1.	С	2.	В	3.	В	4.	А		

EXERCISE - 5 : PART # I

1. 3 **2.** 3 **3.** 2 **4.** 4 **5.** 1 **6.** 4 **7.** 1 **8.** 4 **9.** 1

PART # II

1.3-A	minoben	zoic aci	d 2.	В	3. 4-	Methylbo	enzensulphonic acid 4.	С	5.	В
6.	5	7.	В	8.	D	9.	B,C			



MOCK TEST

1	В	2	С	3	D	4
8	D	9	В	10	С	11
14	В	15	В	16	В	17
21	В	22	В	23	С	24
27	А	28	Е	29	С	30
33	С	34	С	35	С	36
37	CH ₃	-CH ₂ -	$^{4}_{CH}$ – $^{5}_{CH}$	$H_2 - CH_2$	- ⁷ CH ₂ -	- ⁸ CH ₃

А	5	В	6	Α	7	
В	12	В	13	Α		
С	18	В	19	В	20	
A, B, D	25	A, B, D	26	A, B, 0	C, D	
В	31	D	32	В		
$A \rightarrow (Q)$	$B \rightarrow (S$), $C \rightarrow (I$	P), $D \rightarrow ($	R)		

$$CH_{3} - CH_{2} - CH_{3}$$

4-Ethyl octane

- ${}^{1}_{CH_{2}} = {}^{2}_{C} {}^{3}_{CH_{2}} {}^{4}_{CH} {}^{5}_{CH_{3}}$ 38 $CH_3 - CH - CH_3 CH_3$ 2-Isopropyl-4-methylpentene
- 4-Methyl-2-(methyl ethyl) pentene or
- 39 Isopropylidenecyclopentane or 1-methyl ethylidene cyclopentane
- 40 spiro (2.5) octane
- ${}^{1}CH_{2} = {}^{2}CH {}^{3}CH_{2} OCH_{3}$ 41 3-Methoxypropene
- 42

Pentane-2,4-dione

- 43 1,3-cyclobutadiene
- 44 1-cyclohexyl-1-propanone
- 45 5-amino-6(1-methylpropyl) cyclo-hex-2-enol
- 46 Ethyl-2-oxo-cyclo pentane carboxylate
- 4-chloro-1-cyclopentyl pentane-2-one 47
- 1,3-diphenyl-1,4-pentadiene 48

